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computer simulations. The unique properties of polymeric systems are due to the size, topology and interactions of the molecules they are made of.

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Between 1983 and 1985 Michael Rubinstein was a post-doctoral fellow with E. Helfand at AT&T Bell Laboratories in Murray Hill, NJ where he started his research in polymer physics. In 1985 Michael Rubinstein joined Research Laboratories of Eastman Kodak Company in

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Rochester, NY where he worked for 10 years in different areas of polymer theory.

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Thermodynamics of Mixtures 3.
Polymer Solutions Outline . Summary
of Ideal Chains Ideal chains: no
interactions between monomers
separated by many bonds Mean
square end-to-end distance of ideal
linear polymer $R^2 = Nb^2$ Mean square
radius of gyration of ideal linear

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polymer 6 2 R2 Nb g 2 3 / 2 ...

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- Strong dependence of polymer size on environment/solvent conditions suggests a big role of interactions.
- Ideal polymer has no interactions between monomers, except between

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neighbors along the chain. •Just like ideal gas may have all sorts of rotations and vibrations in the molecule, but no interactions between molecules.

Introduction to polymer physics
Lecture 1

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