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11) Which molecule has a dipole moment of zero? A) CHCl_3 B) CH_2Cl_2 C) $\text{ClHC}=\text{CH}_2$ D) *trans*- $\text{ClHC}=\text{CHCl}$ E)

None of these choices. Answer: D

Topic: Molecular geometry, polarity

Section Reference 1: 1.5 and 2.3

Difficulty: Medium 12) Which molecule would have a dipole moment greater than zero? A) BeCl_2 B) BCl_3 C) CO_2 D) H_2O E) CCl_4 Answer: D

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